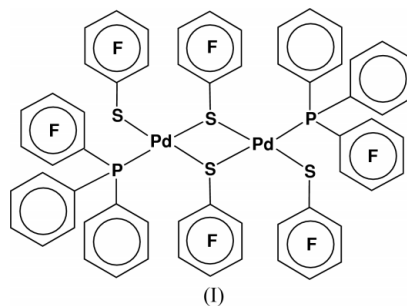


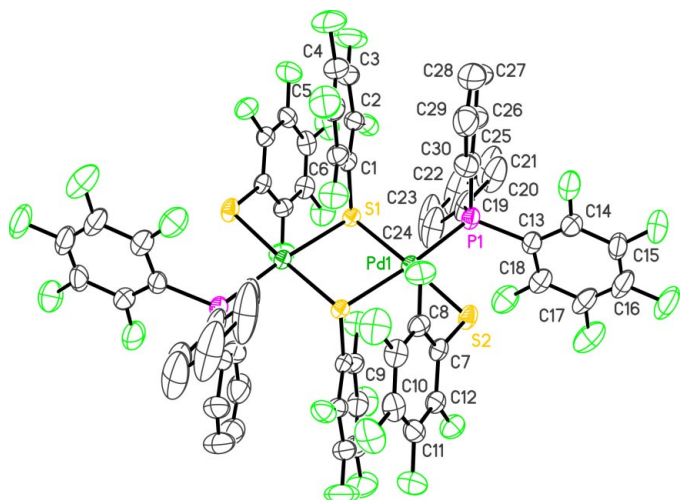
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## Key indicators

Single-crystal X-ray study  
T = 296 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.015 \text{ \AA}$   
R factor = 0.047  
wR factor = 0.111  
Data-to-parameter ratio = 12.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.Bis( $\mu$ -pentafluorobenzenethiolato- $\kappa^2\text{S}:\text{S}$ )-*trans*-bis{(pentafluorobenzenethiolato- $\kappa\text{S}$ )-[(pentafluorophenyl)diphenylphosphine- $\kappa\text{P}$ ]-palladium(II)}In the title compound,  $[\text{Pd}_2(\text{C}_6\text{F}_5\text{S})_4(\text{C}_{18}\text{H}_{10}\text{F}_5\text{P})_2]$ , the molecule lies on an inversion center. The structure is characterized by a nearly planar  $[\text{P},\text{S}-\text{Pd}(\mu_2-\text{S})_2\text{Pd}-\text{S},\text{P}]$  moiety, with *anti* bridging thiolate ligands and terminal ligands in a mutually *trans* arrangement.Received 20 October 2003  
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## Comment

We have a long-standing interest in dinuclear transition metal complexes bearing fluorinated phosphines and thiolates (Claver *et al.*, 1990; García *et al.*, 1993; Arroyo *et al.*, 2000), since these compounds are relevant, for example, to the development of chemical systems involving C–F bond activation (Murai, 1999) or to the study of fluorinated biphasic catalytic processes (Barthel-Rosa & Gladysz, 1999). While working with this class of compounds, we have found that the metathetical reaction of *trans*- $[\text{PdCl}_2\{\text{P}(\text{C}_6\text{H}_5)_2(\text{C}_6\text{F}_5)_2\}]$  with  $\text{Pb}(\text{SC}_6\text{F}_5)_2$  gives rise not only to the expected mononuclear complex *trans*- $[\text{Pd}(\text{SC}_6\text{F}_5)_2\{\text{P}(\text{C}_6\text{H}_5)_2(\text{C}_6\text{F}_5)_2\}]$ , but also to a dinuclear compound, *trans*- $[\{\text{P}(\text{C}_6\text{H}_5)_2(\text{C}_6\text{F}_5)_2\}(\text{SC}_6\text{F}_5)\text{Pd}(\mu_2-\text{SC}_6\text{F}_5)_2\text{Pd}(\text{SC}_6\text{F}_5)\{\text{P}(\text{C}_6\text{H}_5)_2(\text{C}_6\text{F}_5)_2\}]$ , (I), probably through a sterically driven rearrangement of the mononuclear species. We now report the crystal structure of this dinuclear compound.The asymmetric unit of (I) contains one-half of the molecule ( $Z' = \frac{1}{2}$ ) with the Pd atom lying close to an inversion center (Wyckoff position  $4c$  in space group  $C2/c$ ), generating a centrosymmetric dinuclear complex (Fig. 1) including a planar  $[\text{Pd}_2(\mu_2-\text{S})_2]$  core. This symmetry is retained in solution, as reflected in the single signal observed in the  $^{31}\text{P}$  NMR spectrum (see *Experimental*). The bridging thiolate ligands display an *anti* configuration and the coordination sphere is completed by two terminal ligands, a thiolate and a phosphine, in a *trans* arrangement. The resulting coordination geometry for the  $\text{Pd}^{\text{II}}$  ions is very close to square-planar (Table 1): the r.m.s. deviation from the least-squares plane defined by  $\text{S}1/\text{S}1^1/\text{S}2/\text{P}1$



**Figure 1**

The structure of (I), with displacement ellipsoids at the 25% probability level. H atoms have been omitted for clarity. The numbering scheme is given for the asymmetric unit, omitting F atoms for which the number is the same as for the parent C atom.

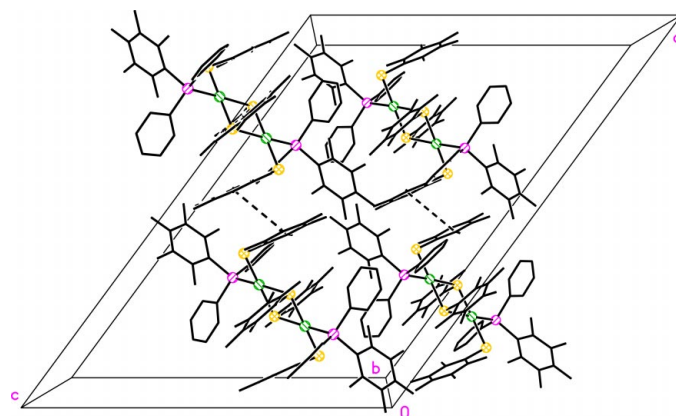
[symmetry code: (i)  $-x + \frac{1}{2}, -y + \frac{3}{2}, -z$ ] is 0.0343 (9) Å, with atom Pd1 deviating from this plane by 0.0074 (7) Å.

The same *trans-anti* conformation was previously observed for closely related complexes, *viz.* with triphenylphosphine as terminal phosphine ligands (Fenn & Segrott, 1972), or with triphenylphosphine and benzenethiolate instead of fluorinated ligands (Nakanishi *et al.*, 1994). In other words, these complexes seem to stabilize this conformation in the solid state, regardless of the F/H substitutions at ligands, *i.e.* regardless of the ligand basicity. Straightforward arguments based on steric considerations can explain the stabilization of this conformation. However, no far-reaching conclusions should be drawn using these observations; for example, the Pt-based complex including phenylmethanethiolate as bridging and terminal ligands and methylphenylphosphine as terminal phosphine ligands displays a *cis-syn* arrangement (Bird *et al.*, 1982). On the other hand, dynamic processes generated by inversion of configuration at the S-bridging atoms can take place, giving several stereoisomeric species in solution, as observed for related Pt-containing complexes (Rivera *et al.*, 2001). Therefore, in the case of (I), the possibility of crystallization of a variety of conformers should not be ruled out.

Finally, weak  $\pi$ - $\pi$  interactions between terminal perfluorothiolate rings are observed in the packing of (I). The closest separation between symmetry-related centroids of rings C7–C12 is 3.7541 (4) Å (symmetry code:  $1 - x, y, \frac{1}{2} - z$ ; Fig. 2).

## Experimental

Equimolar amounts of *trans*-[PdCl<sub>2</sub>{P(C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>(C<sub>6</sub>F<sub>5</sub>)<sub>2</sub>}] (0.226 g) and Pb(SC<sub>6</sub>F<sub>5</sub>)<sub>2</sub> (0.152 g) dissolved in 20 ml of acetone were mixed and kept at room temperature with magnetic stirring for 1 h. After this



**Figure 2**

The packing of (I), viewed down the *b* axis. Four molecules are displayed, with closest  $\pi$ - $\pi$  intermolecular contacts between aromatic rings represented by dashed lines. For clarity, H atoms have been omitted.

period, the resulting white solid (PbCl<sub>2</sub>) was filtered off. The yellow solution was chromatographed through a column using silica gel as the stationary phase (Merck, 0.2–0.5 mm) and a mixture of hexane-dichloromethane (4:1) as eluant, to obtain two yellow fractions and a third colorless fraction. The first fraction contained the mononuclear complex *trans*-[Pd(SC<sub>6</sub>F<sub>5</sub>)<sub>2</sub>{P(C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>(C<sub>6</sub>F<sub>5</sub>)<sub>2</sub>}] (yield 46%). Fraction 2 contained the title compound, (I) (yield 14%). The third fraction contained the free phosphine P(C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>(C<sub>6</sub>F<sub>5</sub>). Single crystals of (I) were obtained by slow evaporation of the solvent under an N<sub>2</sub> flow. Analysis found: C 42.1, H 1.1, S 7.5%; calculated for C<sub>60</sub>H<sub>20</sub>F<sub>30</sub>P<sub>2</sub>Pd<sub>2</sub>S<sub>4</sub>: C 42.0, H 1.2, S 7.5%. MS: *m/z* 1714 (18%); <sup>31</sup>P NMR (in acetone-*d*<sub>6</sub>, referenced to external 85% H<sub>3</sub>PO<sub>4</sub>):  $\delta$  8.26 (s, P<sub>phos</sub>).

## Crystal data

[Pd<sub>2</sub>(C<sub>6</sub>F<sub>5</sub>S)<sub>4</sub>(C<sub>18</sub>H<sub>10</sub>F<sub>5</sub>P)<sub>2</sub>]  
*M<sub>r</sub>* = 1713.74  
 Monoclinic, C2/c  
*a* = 26.807 (3) Å  
*b* = 14.3442 (18) Å  
*c* = 20.3758 (15) Å  
 $\beta$  = 126.362 (6)°  
*V* = 6309.4 (13) Å<sup>3</sup>  
*Z* = 4

*D<sub>x</sub>* = 1.804 Mg m<sup>-3</sup>  
 Mo K $\alpha$  radiation  
 Cell parameters from 36 reflections  
 $\theta$  = 2.8–12.0°  
 $\mu$  = 0.88 mm<sup>-1</sup>  
*T* = 296 (1) K  
 Prism, red  
 0.3 × 0.3 × 0.2 mm

## Data collection

Bruker P4 diffractometer  
 2 $\theta$ / $\omega$  scans  
 Absorption correction:  $\psi$  scan (Fait, 1991)  
*T<sub>min</sub>* = 0.761, *T<sub>max</sub>* = 0.841  
 6476 measured reflections  
 5561 independent reflections  
 3530 reflections with *I* > 2 $\sigma$ (*I*)

*R<sub>int</sub>* = 0.037  
 $\theta_{max}$  = 25.0°  
*h* = -1 → 31  
*k* = -1 → 17  
*l* = -24 → 19  
 3 standard reflections every 97 reflections  
 intensity decay: 4.1%

## Refinement

Refinement on *F*<sup>2</sup>  
*R*[(*F*) > 2 $\sigma$ (*F*)] = 0.047  
*wR*(*F*<sup>2</sup>) = 0.111  
*S* = 1.00  
 5561 reflections  
 442 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0437P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} = 0.001$   
 $\Delta\rho_{max} = 0.44 \text{ e \AA}^{-3}$   
 $\Delta\rho_{min} = -0.31 \text{ e \AA}^{-3}$

**Table 1**

Selected geometric parameters (Å, °).

Pd1—P1	2.2826 (16)	P1—C19	1.834 (7)
Pd1—S2	2.3153 (14)	P1—C13	1.846 (6)
Pd1—S1	2.3586 (13)	S1—C1	1.765 (6)
Pd1—S1 <sup>i</sup>	2.3768 (14)	S2—C7	1.765 (5)
P1—C25	1.810 (8)		
P1—Pd1—S2	89.15 (5)	C19—P1—C13	101.3 (3)
P1—Pd1—S1	92.99 (5)	C25—P1—Pd1	106.9 (2)
S2—Pd1—S1	177.51 (6)	C19—P1—Pd1	113.9 (3)
P1—Pd1—S1 <sup>i</sup>	174.80 (5)	C13—P1—Pd1	117.3 (2)
S2—Pd1—S1 <sup>i</sup>	95.64 (5)	C1—S1—Pd1	107.78 (17)
S1—Pd1—S1 <sup>i</sup>	82.26 (5)	C1—S1—Pd1 <sup>i</sup>	104.12 (18)
C25—P1—C19	110.4 (4)	Pd1—S1—Pd1 <sup>i</sup>	97.74 (5)
C25—P1—C13	106.8 (3)	C7—S2—Pd1	106.48 (17)

Symmetry code: (i)  $-x + \frac{1}{2}, -y + \frac{3}{2}, -z$ .

A refinement carried out using all-free parameters showed the C19—C24 phenyl ring to have a larger than usual  $U_{eq}$  range [ $U_{eq}(\text{max})/U_{eq}(\text{min}) = 2.83$  for this group of six C atoms]. A similarity restraint (Sheldrick, 1998) was then applied to the displacement parameters of this group; atoms closer than 1.5 Å were restrained, with an s.u. value of 0.04 Å<sup>2</sup>, to have the same  $U_{ij}$  components [36 restraints, giving, after refinement,  $U_{eq}(\text{max})/U_{eq}(\text{min}) = 2.39$ ]. H atoms were treated as riding atoms, with  $U_{iso}(\text{H}) = 1.2U_{eq}(\text{parent C})$  and C—H distances constrained to 0.93 Å.

Data collection: *XSCANS* (Fait, 1991); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SHELXTL-Plus* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL-Plus*; molecular graphics: *SHELXTL-Plus*; software used to prepare material for publication: *SHELXTL-Plus*.

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